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_ 5 _ _ _ 1 _ _ _ _ _ (_

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449 14.16 The problem states that the system is at equilibrium, so we simply substitute the equilibrium concentrations into the equilibrium constant expression to calculate K_c .
Step 1: Calculate the concentrations of the components in units of mol/L.

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14.2 What are reducing sugars? Ans. Carbohydrates which reduces Fehling's solution to red precipitate of Cu_2O or Tollen's reagent to metallic Ag are called reducing sugars. All monosaccharides (both aldoses and ketoses) and disaccharides except sucrose are reducing sugars. Thus, D - (+) - glucose,

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D-(-)-fructose, D - (+) - maltose and D - (+) - lactose are reducing sugars.

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The reaction below is first order in [H₂O₂]: $2\text{H}_2\text{O}_2 (\text{l}) \rightarrow 2\text{H}_2\text{O} (\text{l}) + \text{O}_2 (\text{g})$ A solution originally at 0.600 M H₂O₂ is found to be 0.075 M after 54 min. The half-life for this reaction is _____ min. ...
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