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7H 5(NO 2) 3. 4 elements a. How many elements make up this compound? 6 oxygen atoms b. How many oxygen atoms are present in one molecule of C 7H 5(NO 2) 3? 21 atoms c. How many

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atoms in total are present in one molecule of $C_7H_5(NO_2)_3$? 4.2 10^{24} atoms d. How many atoms are present in a sample of 2.0×10^{23} molecules of $C_7H_5(NO_2)_3$? 4.

7 Chemical Formulas and Chemical Compounds

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STUDY. PLAY. Formula mass. The
formula unit of a molecule or ion is the
sum of the average atomic masses of all
atoms represented in its formula.
Percentage composition.

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CHAPTER 7 REVIEW Chemical Formulas
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following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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Compounds

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