

## Modern Chemistry Chapter 5 The Periodic Law Test Answer Key

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### Modern Chemistry Chapter 5 The

About This Chapter The Periodic Law chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with the periodic law. Each of these simple...

### Holt McDougal Modern Chemistry Chapter 5: The Periodic Law ...

Highlights 5.1 - The periodic law states that the physical and chemical properties of the elements are periodic functions of their atomic numbers. - The periodic table is an arrangements of the elements in order of their atomic numbers so that elements with similar properties fall in the same column.

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Modern Chemistry Chapter 5. periodic law. lanthanide. actinide. alkali metals. The physical and chemical properties of the elements are perio.... The 14 elements with the atomic numbers 58-71. The 14 elements with the atomic numbers 90-103. The elements of group 1 of the periodic table.

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HOMEWORK 5-1. 1. Explain how Mendeleev developed the periodic table. 2. Explain the term periodicity. How does the term apply to chemistry? Multiple Choice \_\_\_\_1. Which statement is true? a. The modern periodic table was developed at the First International Congress of Chemists. b. Motion needs to vary uniformly in order to vary periodically. c.

### Modern Chemistry • CHAPTER 5

a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 \_\_\_\_ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. \_\_\_\_ 22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

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