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Compounds Containing
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Naming Compounds Containing Polyatomic Ions

(NH₄)₃PO₄, ammonium phosphate Co
(NO₂)₃, cobalt (III) nitrite

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5.03 Naming Compounds that contain Polyatomic Ions ...

These form compounds that will contain both ionic and covalent bonds. Table of Polyatomic Ions . Rules-Compounds have a first and a last name. 1. Name the cation (+ ion) first (watch for NH_4^+ ...ammonium) 2. Name the anion (- ion)

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second ...that is it (no prefixes...except dichromate) (hydrogen carbonate is also called bicarbonate)

Naming Compounds with Polyatomic Ions - kentchemistry.com

The atoms in a polyatomic ion are usually covalently bonded to one another, and therefore stay together as

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a single, charged unit. Rule 1. The cation is written first in the name; the anion is written second in the name. Rule 2. When the formula unit contains two or more of the same polyatomic ion, that ion is written in parentheses with the subscript written outside the parentheses.

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**Rules for Naming Ionic Compounds
Containing Polyatomic Ions**

Let's make this super easy! This video breaks down what you need to know to pass your next chemistry test on naming compounds with polyatomic ions. Plus I'll...

Naming Compounds with Polyatomic

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Rules for Naming Ionic Compounds
Containing Polyatomic Ions Polyatomic
ions are ions which consist of more than
one atom. For example, nitrate ion, NO_3^- ,
contains one nitrogen atom and three
oxygen atoms. The atoms in a
polyatomic ion are usually covalently
bonded to one another, and therefore

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stay together as a single, charged unit.

Ionic Compounds Containing Polyatomic Ions

How to name ionic compounds containing common polyatomic ions If you're seeing this message, it means we're having trouble loading external resources on our website. If you're

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Polyatomic ions & Common polyatomic ions (article) | Khan ...
Naming Ionic Compounds with Polyatomic Ions DRAFT. 9th - 12th grade. 118 times. Chemistry. 74% average

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accuracy. 9 months ago.

keely_chaisson_12165. 0. Save. Edit.

Edit. ... multiply any subscripts on the polyatomic ion by the subscript you are adding. put brackets around the polyatomic ion before adding a subscript

...

Naming Ionic Compounds with

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Polyatomic Ions Quiz - Quizizz

Naming Ionic Compounds with Polyatomic Ions. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. susan_marcant
TEACHER. Naming ionic compounds with Polyatomic Ions and metals with fixed charges only. No transition metals. Terms in this set (179) Ammonium

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Hydroxide. NH_4OH . Sodium Hydroxide.
 NaOH .

Naming Ionic Compounds with Polyatomic Ions Flashcards ...

Find the name and charge of the polyatomic ion from the list of polyatomic ions. See if the charges are balanced (if they are you're done!) Add

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subscripts (if necessary) so the charge for the entire compound is zero. Use the crisscross method to check your work.
Notes: Don't write the subscript '1'.

How to Write Formulas for Ionic Compounds with Polyatomic Ions

Naming polyatomic ions is harder, but doable. First, name the cation, which is

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just the name of the element. Next, name the anion. This gets trickier. Let's first talk about a polyatomic ionic compound that contains oxygen. If the polyatomic anion contains oxygen, it is called an oxyanion. If it can form more than one form of oxyanion, it gets a suffix of either -ate or -ite. There are some common oxyanions that most

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people memorize.

Naming Ionic Compounds: Simple Binary, Transition Metal ...

Ionic Compounds Containing Polyatomic
Ions Please complete the following table:

Formula of Ionic Compound C (Is ClO

KCH₃CO₂ CaSO₄ NaNO₃ Li₂CO₃

Ba(NO₃)₂ Cs₂CrO₄ NH₄OH 2. 3. 4, 5. 6.

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8. 9. Name of Ionic Compound Sodium
chromate Calcium carbonate Magnesium
nitrate Aluminum sulfate Lithium
phosphate Ammonium chloride Cesium
chlorate Potassium sulfate

Hudson City Schools / Homepage

Naming Compounds Containing
Polyatomic Ions By Debbie McClinton Dr.

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Miriam Douglass Dr. Martin McClinton.
Learners examine a table of common polyatomic ions.

Naming Compounds Containing Polyatomic Ions - Wisc-Online OER

Summarize the rules for naming ionic compounds containing metals that can have multiple charges by placing the

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items below in order with the first step at the top and last at the bottom. ...

Describe the process of naming acids that do not contain polyatomic ions.

Write you answers in order with the first at the top and the last at the bottom.

Che 140 Learn Smart Ch. 3 Flashcards | Quizlet

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Naming Ionic Compounds Using -ite and -ate Some polyatomic anions contain oxygen. These anions are called oxyanions. When an element forms two oxyanions, the one with less oxygen is given a name ending in -ite and the one with more oxygen are given a name that ends in -ate.

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How to Name Ionic Compounds - ThoughtCo

C4.2c Name compounds containing a polyatomic ion. 8. C4.2d Given the name of a compound containing a polyatomic ion, write its formula. Objective 6 Define a polyatomic ion Polyatomic Ion: An ion (either positively or negatively charged) composed of two or more atoms.

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Polyatomic Ions - Michigan State University

To see all my Chemistry videos, check out <http://socratic.org/chemistry> Here's how to write formulas for ionic compounds that contain polyatomic ions. In ord...

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Writing Formulas with Polyatomic Ions - YouTube

One more set of polyatomic ions to know. So we have CrO_4^{2-} , which is called Chromate. And if we have two Chromiums, so $\text{Cr}_2\text{O}_7^{2-}$ this is called Dichromate. Next, $\text{C}_2\text{O}_4^{2-}$ is called the Oxalate ion. and we have O_2^{2-} , two

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minus is called Peroxide.

Common polyatomic ions (video) | Khan Academy

When you write the formula for an ionic compound, remember that the positive ion is always listed first. When there are two or more polyatomic ions in a formula, enclose the polyatomic ion in

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parentheses. Write down the information you have for the charges of the component ions and balance them to answer the problem.

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