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-Study of the QUANTITY

relationships in a

CHEMICAL REACTION

-Based on BALANCED

equations  $2 \text{ Mg} + \text{O}_2 \rightarrow 2$

$\text{MgO}$  1. Particles 2

atoms Mg 2 atoms O 2

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atoms Mg 2 atoms O 2  
atoms Mg 1 molecule  
O<sub>2</sub> 2 molecule MgO 10  
atoms Mg 5 molecule  
O<sub>2</sub> 10 molecule MgO 2  
( $6.02 \times 10^{23}$ ) 1 ( $6.02 \times 10^{23}$ ) atoms Mg  
molecule O<sub>2</sub> 2 ( $6.02 \times 10^{23}$ ) molecule MgO  
2.

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Stoichiometry Problem  
Guidelines: 1. Balance  
the

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atoms contain equal  
numbers of positive c  
negative move to  
create unbalanced  
charges. and When  
atoms lose electrons

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they become charged ions. Like charges and opposite charges Friction, one added -hha+ o Rubber material of a balloon has a greater attraction for electrons than hair.

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of the UUUD based on  
 $2 \text{ Mg} + \text{O}_2 \rightarrow 2 \text{ MgO}$  Name  
relationships in a  
equations cc L, ch  
cpnöducc- The Ex.  
Problem: in a give the  
for involved in the  
When elemental  
aluminum reacts with

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elemental iodine,  
aluminum iodide is  
produced. mole ratios:  
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