

## Word And Skeleton Equations Answers

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1. Solid mercury(II) oxide breaks down when heated, forming the elements mercury and oxygen. 2. Sodium metal reacts with water vapor in air to form solid sodium hydroxide and hydrogen. 3. certain pollutants in the air react with water vapor to form acids. For example, sulfur trioxide reacts with water vapor to form sulfuric acid. 4. Solid calcium carbonate is commonly used in antacids because ...

### Chemical Reactions = Word Equations, a Skeleton equation ...

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### Word And Skeleton Equations Answers

Word Chemical Equations Answer Sheet Word Equations Worksheet - Solutions. Write the skeleton equations for each of the following chemical reactions: 1) When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.

### Word Chemical Equations Answer Sheet

word equation skeleton equation 7. Gold (III) oxide decomposes completely at high temperatures to produce metallic gold and oxygen gas. word equation skeleton equation 8. For each of the following chemical formulas, determine if it is ionic or covalent and write the correct name for the compound represented. a) N

### Worksheet 1- Word and Skeleton Equations ACh

Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead.  $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$  Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas.  $2AlBr_3 + 3Cl_2 \rightarrow 2AlCl_3 + 3Br_2$  Single Replacement 3.

### Word Equations Answer Key Zinc and Lead (II) nitrate react ...

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word equation: methane + oxygen → water + carbon dioxide 2. Write a skeleton equation: A skeleton equation replaces the names of the reactants and products in a word equation with formulas. However, it does not show the correct proportions in which the reactants will actually combine and the products will be produced. A skeleton equation is not balanced. skeleton equation: CH<sub>4</sub> + O<sub>2</sub>

### **Section Name Date 4.3 Chemical Equations**

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